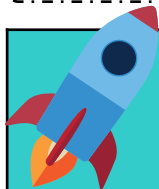


Water is an example of a **chemical compound** - when two or more elements join together to form molecules.

Water is 2 Hydrogen (H) atoms + 1 Oxygen (O) atom = **H₂O**

Knowledge Organiser Changes of Materials

A '**mixture**' in a scientific sense, can always be broken down into its component parts.



ROCKET WORDS

Learn these words and their definitions.

Key Word	Definition
separate	To split or divide a substance into its distinct elements
solution	A mixture of two substances, the solute and the solvent
solute	A substance that is dissolved in liquid.
solvent	A substance that dissolves a solute, such as water.
irreversible	Impossible to change back to a previous condition or state.
compound	A substance formed when two or more chemical elements are bonded together.
physical change	A change in material in which no new substances are formed
chemical change	A change that results in the creation of few chemical substances.

5 ways to compare a **physical** and **chemical** change.

Property	Physical Change	Chemical Change
Explanation	Molecules are rearranged but the actual type of molecules stay the same.	The type and make-up of the molecules is changed and a new substance is formed.
Change	A temporary change that is easily reversed, and no new substance is formed.	A permanent change that is irreversible, with a new substance always being formed.
Energy	No energy is produced, and very little or no energy is absorbed.	Energy is produced, in the form of light or heat (for example) and energy is also absorbed.
Effects	Only has an effect on physical properties of a substance or object i.e. shape, size.	Changes both physical and chemical properties of a substance or object.
Examples	Freezing or boiling water, melting wax	Burning wood, eating food, rusting of metal.

Lesson Sequence

- 1 Understand the actions of filtering, sieving and evaporating
- 2 Discover how to dissolve something in a solution
- 3 Understand that some changes are not reversible
- 4 Explore permanent chemical change
- 5 Know the difference between physical and chemical change
- 6 Know the difference between elements, compounds and mixtures

Filtering



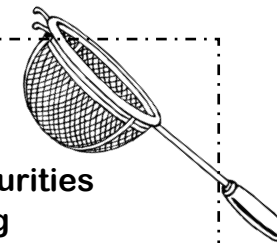
- Brewing coffee
- Cleaning a swimming pool
- Vacuum Cleaning

Evaporating



- Body sweat
- The water cycle
- Salt / crystal extraction

Sieving



- Removing impurities during cooking
- Sieving sand during building
- Mining for minerals

